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B. Tech
CPCH 7307

Sixth Semester Examination – 2008

MASS TRANSFER – II

Full Marks – 70

Time : 3 Hours

Answer Question No. 1 which is compulsory
and any **five** from the rest.

The figures in the right-hand margin
indicate marks.

1. Answer the following questions : 2×10
- (a) What do you mean by constant drying conditions ?
- (b) Suggest suitable dryers for drying the following materials :
- (i) Milk powder



- (ii) Wet lumpy solids
- (iii) Free flowing materials.
- (c) State whether fluidized bed dryers can be used for heat sensitive materials or not.
- (d) State two unit operations involving simultaneous heat and mass transfer.
- (e) Define selectivity.
- (f) Can two tie lines intersect within the two-phase region of a liquid-liquid equilibrium diagram ? Explain qualitatively.
- (g) What are the advantages and problems of carrying out extraction of a solid at an elevated temperature ?
- (h) Select suitable adsorbents for the following purposes :
- (i) Drying of air for pneumatic instruments.

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Contd.

(ii) Removal of phenolic contaminants from wastewater.

(i) Name different methods of generating supersaturation.

(j) Explain Caking of crystals.

2. It is desired to dry a certain type of fiber board in sheets 0.131 m by 0.162 m by 0.071 m from 58% to 5% moisture (wet basis) content. Initially from laboratory test data with this fiber board, the rate of drying at constant rate period was found to be $8.9 \text{ kg/m}^2\text{hr}$. The critical moisture content was 24.9% and the equilibrium moisture content was 1%. The fiber board is to be dried from one side only and has a bone-dry density of 210 kg/m^3 . Determine the time required for drying. The falling rate may be assumed linear.

10

3. (a) Briefly explain about the selection criteria for solvent to be used for liquid-liquid extraction. 5

(b) What is a triangular diagram and how a ternary system is represented on it? 5

4. 2000 kg of waxed paper per day are to be dewaxed in a continuous counter current extraction system which contains a number of ideal stages by using kerosene as solvent. The waxed paper contains, by weight 25% paraffin wax and 75% paper pulp. The extracted pulp is put through a dryer to evaporated the kerosene. The pulp, which retains the unextracted wax after evaporation, must contain over 0.2 kg of wax per 100 kg of wax-free pulp. The kerosene used for extraction contains 0.05 kg of wax per 100 kg of wax-free kerosene. Experiments show that

the pulp retains 2.0 kg of kerosene per kg of kerosene and wax-free pulp as it is transferred from cell to cell. The extract from the battery is to contain 5 kg of wax per 100 kg of wax-free kerosene. How many stages are required ? 10

5. A hot solution containing 5000 kg of Na_2CO_3 and water with a concentration of 25% by weight. Na_2CO_3 is cooled to 20 °C and crystals of $\text{Na}_2\text{CO}_3 \cdot 10 \text{H}_2\text{O}$ are precipitated. At 20 °C, the solubility is 21.5 kg anhydrous Na_2CO_3 per 100 kg of total water. Calculate the yield of hydrated Na_2CO_3 crystals obtained if 5% of the original water in the system evaporates on cooling. Also calculate the quantity of mother liquor. 10

6. (a) Discuss the different criteria for selection of adsorbents. 5

(b) Explain about different methods of bed regeneration used in adsorption. 5

7. A 2500 kg batch of pyridine-water solution, 50% pyridine, is to be extracted once with 2200 kg of chlorobenzene. Determine the concentration of pyridine in the final raffinate and extract.

Equilibrium tie-line data for the system water-chlorobenzene-pyridine at 25 °C are given below :

Pyridine	Chloro-benzene	Water	Pyridine	Chloro-benzene	Water
0	99.95	0.05	0	0.08	99.92
11.05	88.28	0.67	5.02	0.16	94.82
18.95	79.9	1.15	11.05	0.24	88.71
24.1	74.28	1.62	18.9	0.38	80.72
28.6	69.15	2.25	25.5	0.58	73.92
31.55	65.58	2.87	36.1	1.85	62.05
35.05	61	3.95	44.95	4.18	50.87
40.6	53	6.4	53.2	8.9	37.9
49	37.8	13.2	49	37.8	13.2

8. Write short notes on any *four*: 2.5 × 4

(a) Solutropic system

(b) Spray dryer

(c) Selectivity

(d) Super critical extraction

(e) Tray dryer

(f) Heat of adsorption.
